Low-pressure nitrogen afterglows with oxygen addition

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Nitrogen discharges and their afterglows are interesting for a number of applications, such as surface modification of materials, atmospheric plasma chemistry, pollution abatement, or plasma sterilization. The nitrogen afterglow may be divided into two regions. In the early (pink) afterglow, there is a raise in the concentrations of several species after an initial decay [1] and a strong emission of the first negative system. The kinetics of the N₂(*B*) in this zone is governed by reactions involving vibrationally excited molecules, N₂(*X*,v), as well as N₂(*A*) and N₂(*C*) molecules and nitrogen atoms. The late (yellow) afterglow is dominated by the 3-body recombination of N atoms, forming predominantly N₂(*B*,v=11), and the first positive system emission.

Understanding the elementary processes ruling the concentrations of the different active species is a crucial step in order to optimize any specific application. Measurements of nitrogen and oxygen atomic concentrations in the afterglow, of the N₂(A) density in the late afterglow, and of the modification of the emission intensities from the 1⁻ and NO β systems under oxygen addition, have been reported very recently [2,3]. In this work, the model developed in [1] to study the kinetics of the nitrogen afterglow is extended to investigate the influence of small amounts of oxygen added into a nitrogen afterglow and interpret the results in [2,3]. It is shown that a very important effect is the destruction of vibrationally excited molecules by O atoms, in V-T collisions and/or in the NO formation reaction N₂(X,v \geq 13)+O \rightarrow NO+N, which subsequently affect the chain of reactions taking place along the pink afterglow.

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[2] N. Kang, M. Lee, A. Ricard, and S. Oh, "Effect of controlled O_2 impurities on N_2 afterglows of RF discharges," *Curr. Appl. Phys.* **12** (2012) 1448–1453.

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